

[$\pi 6_s + \pi 2_s$] Cycloadditions Catalysed by the $\text{TiCl}_4\text{-Et}_2\text{AlCl}$ System

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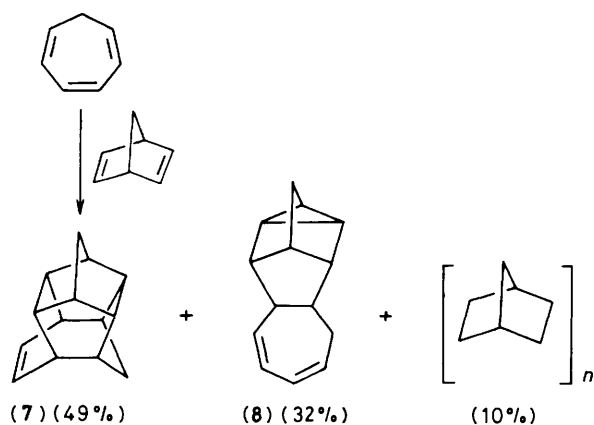
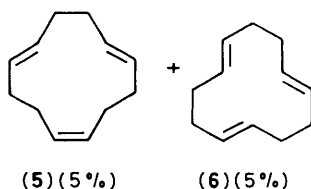
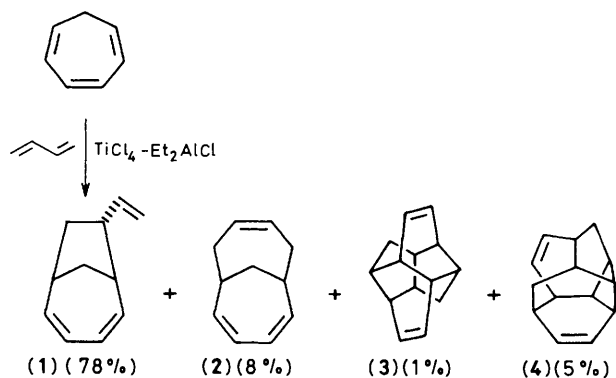
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The catalyst system $\text{TiCl}_4\text{-Et}_2\text{AlCl}$ induces [$\pi 6_s + \pi 2_s$] cycloadditions of cycloheptatriene to buta-1,3-diene, norbornadiene, and acetylenes.

There are only a few reported examples of [$\pi 6_s + \pi 2_s$] cycloadditions.¹⁻⁵ We have recently reported that low-valent titanium complexes, $(\eta^6\text{-C}_6\text{H}_6)\text{Ti}^{\text{II}}(\text{AlCl}_4)_2$ in combination with Et_2AlCl and the Ziegler catalyst $\text{TiCl}_4\text{-Et}_2\text{AlCl}$, induced a

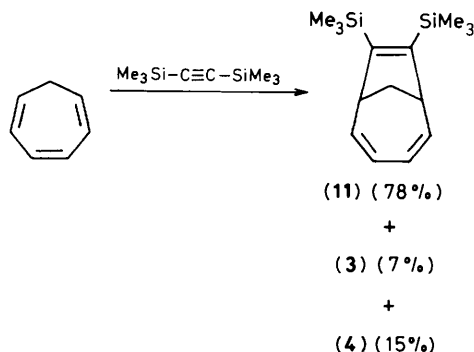
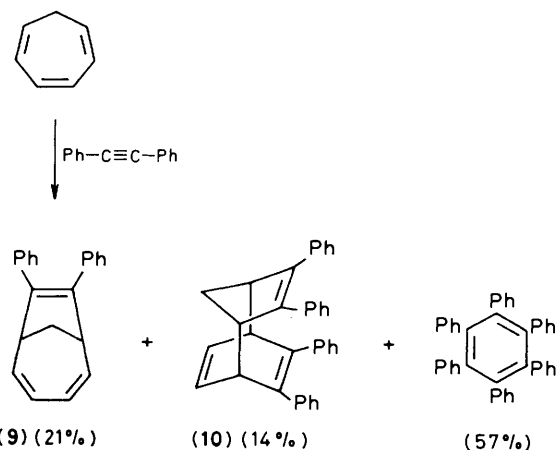
tandem [$\pi 6_s + \pi 2_s$]-[$\pi 4_s + \pi 2_s$] cyclodimerization of cycloheptatriene yielding two pentacyclic tetradecadienes.^{6,7} We now report on the use of the $\text{TiCl}_4\text{-Et}_2\text{AlCl}$ catalyst for cross-additions of cycloheptatriene to other trienophiles.



Buta-1,3-diene and cycloheptatriene (molar ratio $C_4H_6 : C_7H_8 : Et_2AlCl : TiCl_4 = 100 : 100 : 20 : 1$, for experimental details see ref. 7) afford mainly 7-endo-vinylbicyclo[4.2.1]nona-2,4-diene (1), a product of the $[\pi_6s + \pi_2s]$ reaction, accompanied by bicyclo[4.4.1]undeca-2,4,8-triene (2), the cycloheptatriene dimers (3) and (4), and the butadiene cyclo-trimers (5) and (6). Isoprene reacts similarly, giving a mixture of isomeric cycloaddition products, while (*E,E*)-hexa-2,4-diene and cyclo-octa-1,3-diene are unreactive.

Addition of cycloheptatriene to norbornadiene proceeds in two ways: the $[\pi_6s + \pi_2s]$ cycloaddition, followed by an intramolecular Diels-Alder reaction, affords the hexacyclic compound (7) as the main product, while the minor adduct (8) arises via a $[\pi_2s + \pi_2s + \pi_2s]$ mechanism. The formation of (7) could alternatively be visualized as occurring by a $[\pi_2s + \pi_2s + \pi_2s + \pi_2s]$ cycloaddition.

The catalysed addition of cycloheptatriene to acetylenes competes with cyclotrimerization of the latter compounds. But-2-yne and diphenylacetylene trimerize very rapidly, although in the latter case the $[\pi_6s + \pi_2s]$ cycloaddition also takes place to give the bicyclononatriene (9).⁴ This reacts further with diphenylacetylene to give the Diels-Alder adduct (10). Bis(trimethylsilyl)acetylene appeared to be the best alkyne π_2s component as it did not trimerize and added



cleanly to cycloheptatriene, yielding the bicyclic compound (11).[†]

The choice of possible trienophiles for the $[\pi_6s + \pi_2s]$ reaction is limited by two factors: first, the trienophile should not contain reducible groups (carbonyl, halogen, nitro, etc.) and, secondly, it should be able to complex weakly to the catalyst, preferentially through an auxiliary π -orbital.

Received, 13th April 1983; Com. 464

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[†] All products were characterized by their i.r., ¹H and ¹³C n.m.r., and mass spectra. Compounds (1), (2), (8), and (11) are liquids; m.p.s: (7), 168–169 °C; (9), 53–55 °C; (10), 175–177 °C; ¹³C n.m.r. data (δ , p.p.m.): (1): 141.0 (d), 139.4 (d), 135.1 (d), 126.1 (d), 123.3 (d), 114.5 (t), 61.3 (d), 46.4 (d), 41.5 (t), 36.8 (d), and 33.5 (t); (2): 138.24 (d, 2C), 132.11 (d, 2C), 124.79 (d, 2C), 38.30 (d, 2C), 37.00 (t, 2C), and 32.15 (t, 1C); (7): 130.8 (d, 2C), 53.7 (d, 2C), 47.0 (d, 2C), 43.1 (d, 2C), 42.9 (d, 2C), 41.8 (d, 2C), 39.5 (t, 1C), and 35.4 (t, 1C); (8): 137.0 (d), 133.8 (d), 128.1 (d), 125.6 (d), 56.0 (d), 49.9 (d), 49.2 (d), 42.6 (d), 40.0 (d), 31.4 (t), 30.7 (t), 14.3 (d), 11.6 (d), and 10.8 (d); (9): 139.23 (d, 2C), 136.65 (s, 2C), 135.93 (s, 2C), 129.11 (d, 4C), 127.97 (d, 4C), 126.51 (d, 2C), 124.62 (d, 2C), 49.23 (d, 2C), and 30.50 (t, 1C); (11): 146.5 (s, 2C), 137.8 (d, 2C), 123.9 (d, 2C), 51.9 (d, 2C), 30.1 (t, 1C), and 1.79 (q, 6C).